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~~Experiment 12.2 A~~

~~Simple Galvanic Cell  
(Berean Builders)~~

*Galvanic Cell Battery*

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Experiments In A

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to Chem lab, experiment

3 question 1

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Basic Chemistry Lab

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12. Inert gasses are arranged in what part of the Periodic Table? Far right group or column.

PART III - Chemical Reactions, Reactants and Products. 13. A liquid that forms a gas when heated is an example of a chemical

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Answers. False 14.

When Vinegar (acetic acid  $\text{CH}_3\text{COOH}$ ) reacts with baking soda ( $\text{NaHCO}_3$ ), one of the products is Carbon Dioxide ( $\text{CO}_2$ ). True 15.

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part of the Periodic

Table? Far right group

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Experiment 12 1.

Introduction: The objective of this experiment was to measure the physical properties of pressure, temperature, and volume for a gaseous substance. Another objective was to determine the molar mass of the volatile liquid. Volatile liquids



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are molecular  
substances that have  
very low boiling points  
and small masses. .  
Molar masses are  
among the first ...

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molecule on the reactant  
side. Because we added  
HCl and NaOH in steps

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1-4, the colours changed back to their original shades to maintain equilibrium. 7)  $\text{BaCr}_2\text{O}_7$  is much more soluble than  $\text{BaCrO}_4$ , as the  $\text{K}_2\text{Cr}_2\text{O}_7$  solution was darker in colour when mixed with  $\text{Ba}(\text{NO}_3)_2$ . 8) No precipitate formed in step 7 of Part IV because  $\text{H}^+$  in  $\text{HCl}$  was added, shifting the ?

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Keith began writing for the Fool in 2012 and focuses primarily on healthcare investing topics. His background includes serving in management and consulting for the healthcare technology, health ...

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