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dissolving process. Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid. b.

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13 Ions in Aqueous Solutions and Colligative Properties

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questions in the space provided. 1. Describe the errors made by the following
students in making molar solutions. 3. James needs a 0.600 M solution of KCl. He
measures out 0.600 g of KCl and adds 1 L of water to the solid. James made several
errors.

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water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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